

Condition Air into the Human Comfort Zone

▼ Introduction

Humans generally feel comfortable between temperatures of 22°C to 27°C and a relative humidity of 40% to 60%.

In this application, air at 35°C and 60% relative humidity will be conditioned into the human comfort zone, with the thermodynamic process plotted on a psychrometric chart. To do this, we will

- first cool the air to 14°C (this removes some of the water from the air),
- and then heat the air to 24°C.

Additionally, we will calculate

- the heat and mass of water removed in the cooling phase,
- and the heat added in the heating phase.

▼ Plot the Comfort Zone on the Psychrometric Chart

```
> restart :
with( ThermophysicalData ) : with( plots ) : with( Units[ Standard ] ) :
```

Functions for the lower and upper bounds of the human comfort zone.

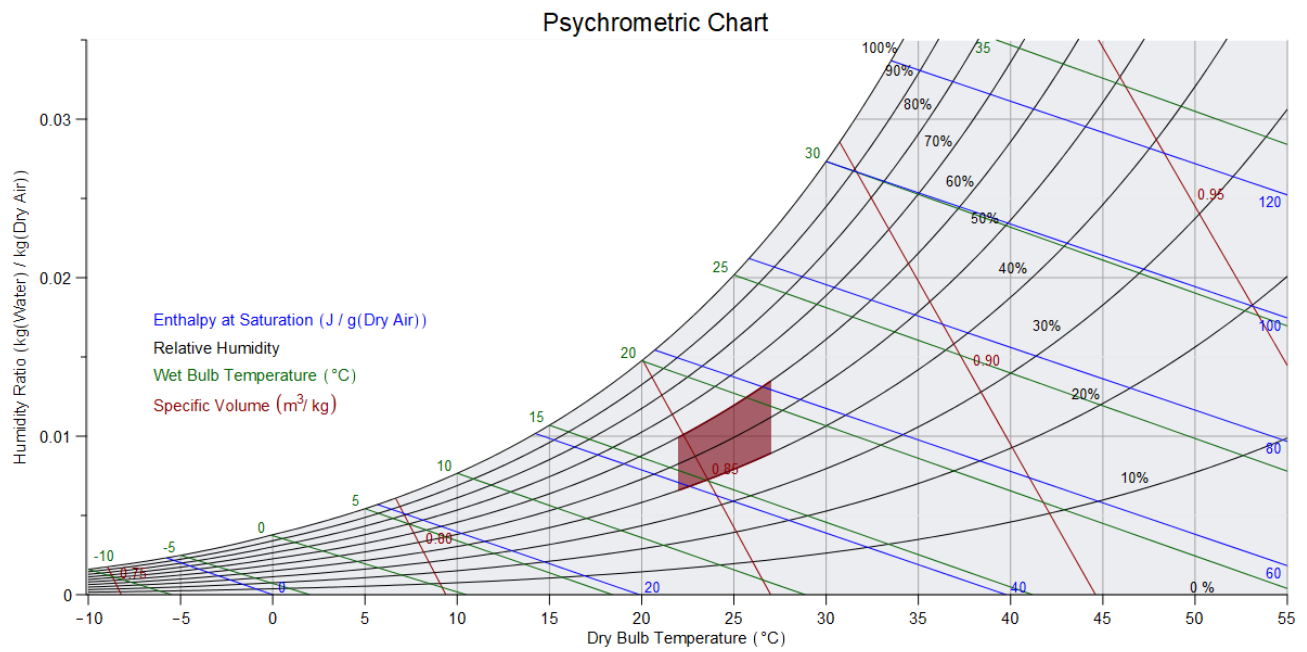
```
> lower := T→Property( humidityratio, HumidAir, P = 101325, Tdb = T, R = 0.4 ) :
> upper := T→Property( humidityratio, HumidAir, P = 101325, Tdb = T, R = 0.6 ) :
```

Shade the human comfort zone between 22°C and 27°C.

```
> comfortZone := shadebetween( lower, upper, 273.15 + 22 .. 273.15 + 27 ) :
```

Plot the human comfort zone on a psychrometric chart.

```
> display( PsychrometricChart( ), comfortZone )
```



▼ Plotting the Thermodynamic Cycle

Initially the air is at a temperature of 35 °C at a relative humidity of 60%

$$> T_1 := 35 \text{ degC} + 273.15 \text{ K} :$$

$$> hr_1 := \text{Property}(\text{humidityratio}, \text{HumidAir}, \text{Tdb} = T_1, "P" = 101325 \text{ Pa}, "R" = 0.6)$$

$$hr_1 := 0.02154666081$$

(3.1)

Then, we cool the air, and calculate the temperature at saturation (that is, the temperature at which the relative humidity is 1).

$$> T_2 := \text{Property}(\text{Tdb}, \text{HumidAir}, R = 1, P = 101325 \text{ Pa}, \text{humidityratio} = hr_1)$$

$$T_2 := 299.2227624 \text{ K}$$

(3.2)

$$> hr_2 := hr_1 :$$

We continue cooling along the saturation line until we reach 14 °C (in this process, water condenses out of the air).

$$> T_3 := 14 \text{ degC} + 273.15 \text{ K} :$$

$$> hr_3 := \text{Property}(\text{humidityratio}, \text{HumidAir}, \text{Tdb} = T_3, P = 101325 \text{ Pa}, R = 1)$$

$$hr_3 := 0.01001332280$$

(3.3)

Now we heat the air until it reaches 24 °C.

$$> T_4 := 273.15 \text{ K} + 24 \text{ degC} :$$

$$> hr_4 := hr_3 :$$

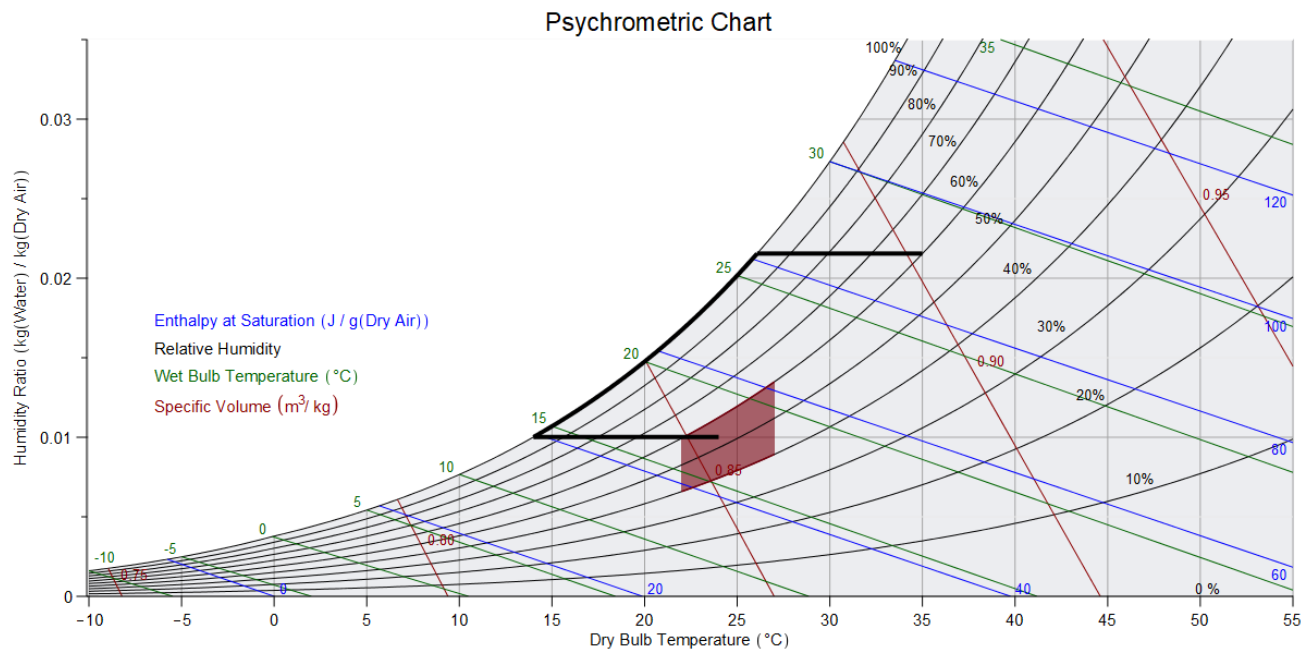
Hence the entire thermodynamic cycle can then be plotted.

$$> \text{route1} := \text{pointplot}(\text{convert}(\text{[[}[T_1, hr_1], [T_2, hr_2]], \text{unit_free}}), \text{connect} = \text{true}, \text{thickness}$$

```

= 4) :
route2 := pointplot( convert( [ [ T3, hr3 ], [ T4, hr4 ] ], unit_free ), connect = true, thickness
= 4) :
> satLine := plot( Property( "humidityratio", HumidAir, Tdb = T, P = 101325, R = 1 ), T = T2..T3,
color = black, thickness = 4) :
> display( PsychrometricChart( ), comfortZone, route1, route2, satLine)

```



▼ Heat Changes and Water Removed over the Thermodynamic Cycle

Water removed in the cooling process (in kg water per kg dry air)

$$> hr_2 - hr_3$$

$$0.01153333801 \quad (4.1)$$

Heat removed in the cooling process (in J kg⁻¹)

$$> h_1 := \text{Property}(\text{enthalpyperdryair}, \text{HumidAir}, \text{Tdb} = T_1, P = 101325, R = 0.6) :$$

$$> h_3 := \text{Property}(\text{enthalpyperdryair}, \text{HumidAir}, \text{Tdb} = T_3, P = 101325, R = 1) :$$

$$> h_1 - h_3$$

$$51101.11785 \frac{\text{J}}{\text{kg}} \quad (4.2)$$

Heat added in the heating process (in J kg⁻¹)

$$> h_4 := \text{Property}(\text{enthalpyperdryair}, \text{HumidAir}, \text{Tdb} = T_4, P = 101325, R = 0.5) :$$

$$> h_4 - h_3$$

$$(4.3)$$

$$8537.89452 \frac{\text{J}}{\text{kg}}$$

(4.3)